

Prob. 2.7

(a) Relation for derivatives:

$$\left(\frac{\partial T}{\partial \mathcal{L}}\right)_s = \frac{\left(\frac{\partial \mathcal{S}}{\partial \mathcal{L}}\right)_T}{\left(\frac{\partial \mathcal{S}}{\partial T}\right)_L} \quad (1)$$

Second law along with heat content as mass M specific heat c temperature change dT :

$$dS = dQ/T, \quad dQ = McdT = TdS \Rightarrow \left(\frac{\partial \mathcal{S}}{\partial \mathcal{L}}\right)_L = \frac{Mc}{T} \quad (2)$$

Substituting (2) into (1);

$$\boxed{\left(\frac{\partial T}{\partial \mathcal{L}}\right)_s = \frac{-T}{Mc} \left(\frac{\partial \mathcal{S}}{\partial \mathcal{L}}\right)_T} \quad (3)$$

(b) First and second law, with $dU = 0$ for an ideal rubber:

$$dU = 0 = dQ + dW = TdS + FdL \Rightarrow \frac{\partial \mathcal{S}}{\partial \mathcal{L}} = -\frac{F}{T}$$

Using this in Eq. (3) of the previous problem:

$$\frac{\partial T}{\partial \mathcal{L}} = \frac{F}{Mc} \quad (4)$$

Extension ratio: $\lambda = L/L_0 \Rightarrow dL = L_0 d\lambda$. Using this in Eq. (4):

$$\frac{\partial T}{\partial \lambda} = \frac{FL_0}{Mc} = \frac{FAL}{A_0 Mc} \quad (5)$$

Now substituting

$$\frac{AL}{M} = \frac{V}{M} = \frac{1}{\rho}$$

where V is volume and ρ is density, along with the engineering stress $\sigma = F/A_0$ into (5):

$$\frac{\partial T}{\partial \lambda} = \frac{\sigma}{\rho c} \Rightarrow dT = \frac{\sigma}{\rho c} d\lambda$$