

LECTURE 12: VAN DER WAALS FORCES AT WORK: GECKO FEET ADHESION

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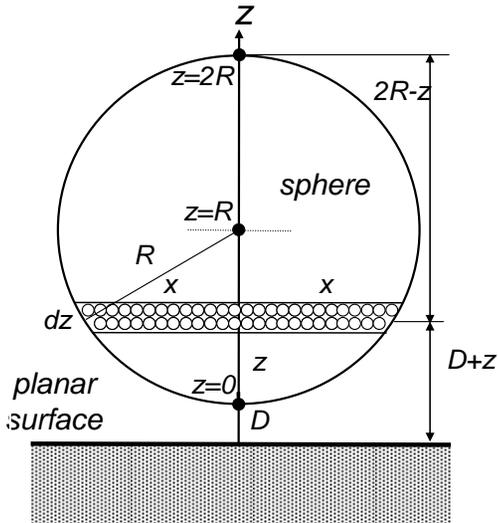
Objectives: To understand how weak van der Waals force can lead to enormous, reversible adhesion

Readings: K. Autumn, *American Scientist*, 94 124 **2006** and Tian, et al. *PNAS* **2006** 103, 51, 19320.

Multimedia : K. Autumn, Discovery Channel movie.

COLLOIDS AND INTERPARTICLE FORCES

-**Definitions;** Colloids, colloidal dispersion, colloidal inks; percolation



$$W(D)_{\text{SPHERE-SFC}} = \frac{-4\pi^2 A \rho^2 R}{(n-2)(n-3)(n-4)(n-5)D^{n-5}}$$

$$W(D)_{\text{SPHERE-SFC}}(\text{VDW}, n=6) = \frac{-\pi^2 A \rho^2 R}{6D}$$

$w(r) \sim r^{-6}$, $W(D)_{\text{MOL-SFC}} \sim D^{-3}$, $W(D)_{\text{SPHERE-SFC}} \sim D^{-1}$
 "Hamaker Constant":

$A = \pi^2 A \rho^2$ (interactions between the same material)

$A = \pi^2 A \rho_1 \rho_2$ (interactions between different materials)

$A \sim 10^{-19} \text{ J}$

$$W(D)_{\text{SPHERE-SFC}}(\text{VDW}, n=6) = \frac{-AR}{6D}$$

$$F(D)_{\text{SPHERE-SFC}}(\text{VDW}, n=6) = -\frac{\partial W(D)}{\partial D} = \frac{-AR}{6D^2}$$

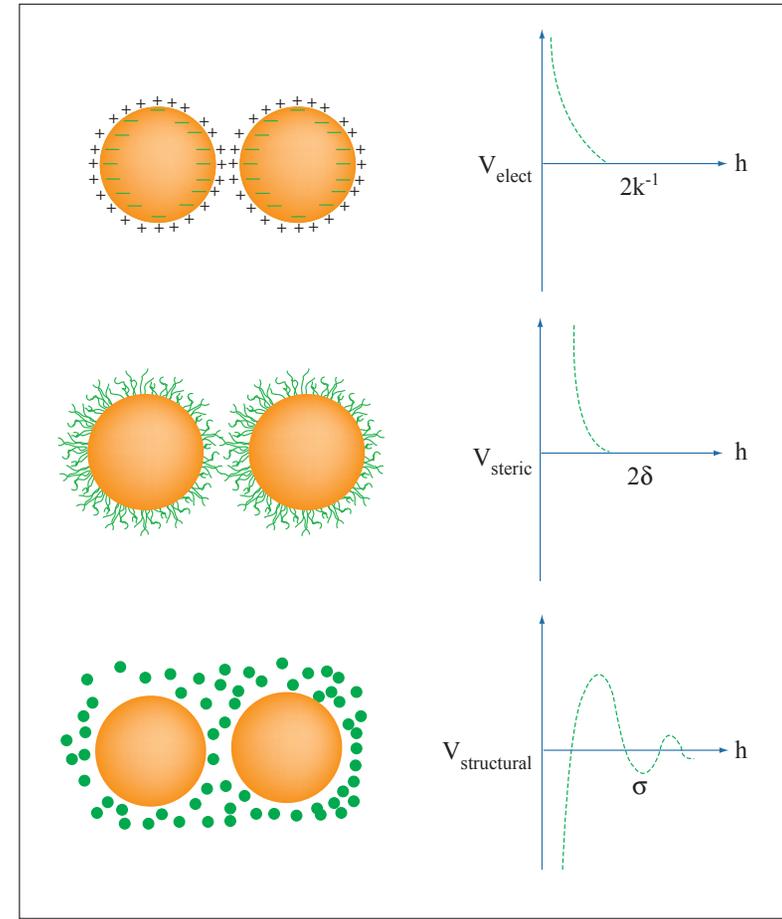


Figure by MIT OCW.
 After Lewis. *J Am Ceram Soc* 83 no. 10 (2000): 2341-59.

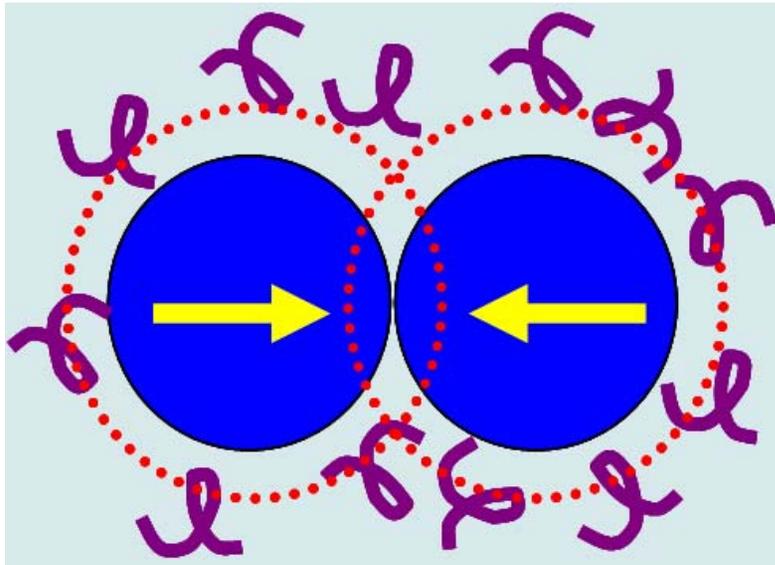
-Analytical formulas for VDW interactions for other geometries

-Colloidal stability, other long range forces; electrostatic double layer, steric, electrosteric, structural, depletion

$$W(D) = \underbrace{W(D)_{\text{VDW}} + W(D)_{\text{ELECTROSTATIC}}}_{\text{DLVO Theory}} + W(D)_{\text{STERIC}} + W(D)_{\text{STRUCTURAL}} - W(D)_{\text{DEPLETION}}$$

COLLOIDAL STABILITY: EFFECT ON DISPERSION

"Depletion Interaction" : For entropic reasons the chains avoid the space between two close particles, or between a particle and a planar wall, and create an effective **attraction** among the colloid particles.



Dispersed state : repulsive energy barrier $\gg k_B T$

Weakly Flocculated : well depth $\sim 2-20 k_B T$

Strongly Flocculated : deep primary minimum

-e.g. Dispersion of nanotubes

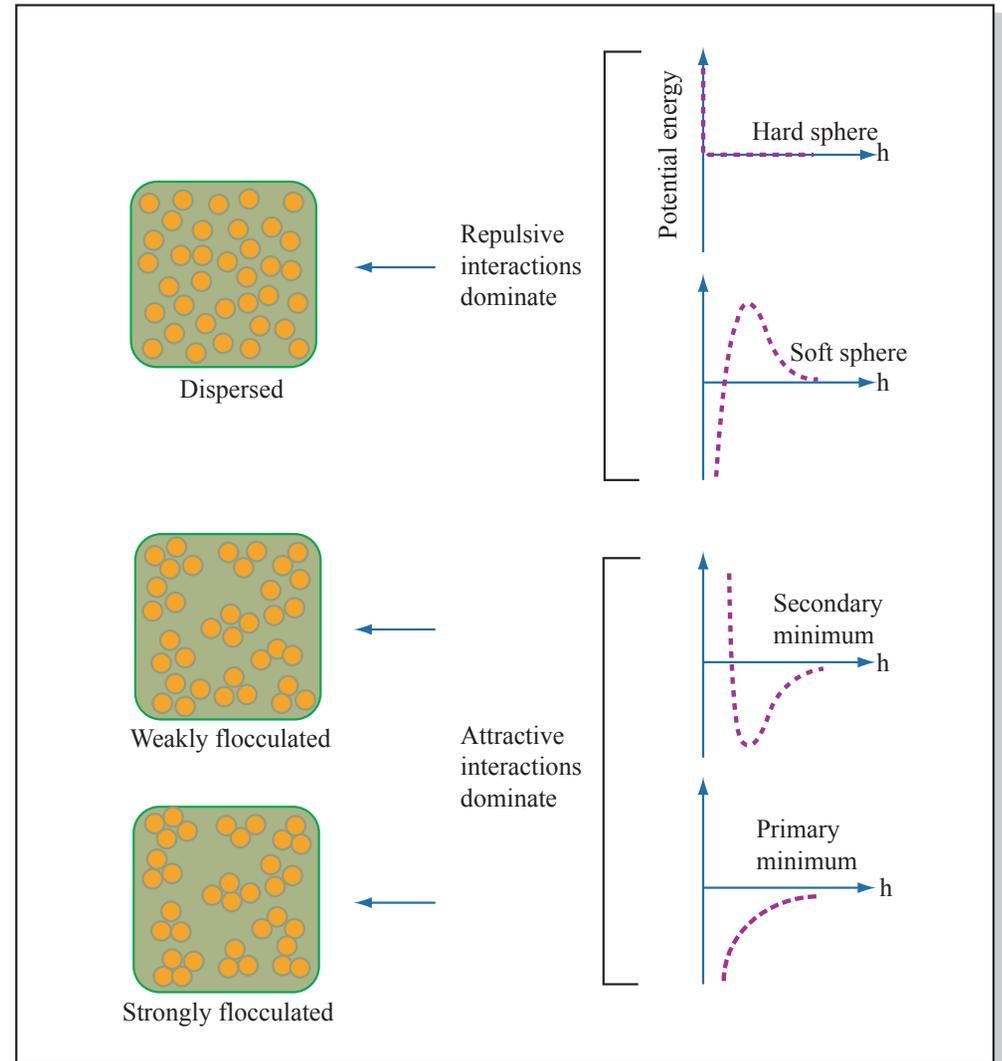


Figure by MIT OCW.
After Lewis. *J Am Ceram Soc* 83 no. 10 (2000): 2341-59.

GECKO ADHESION - "STICKY FEET" (From K. Autumn, et al. *American Scientist*, 2006, 124)



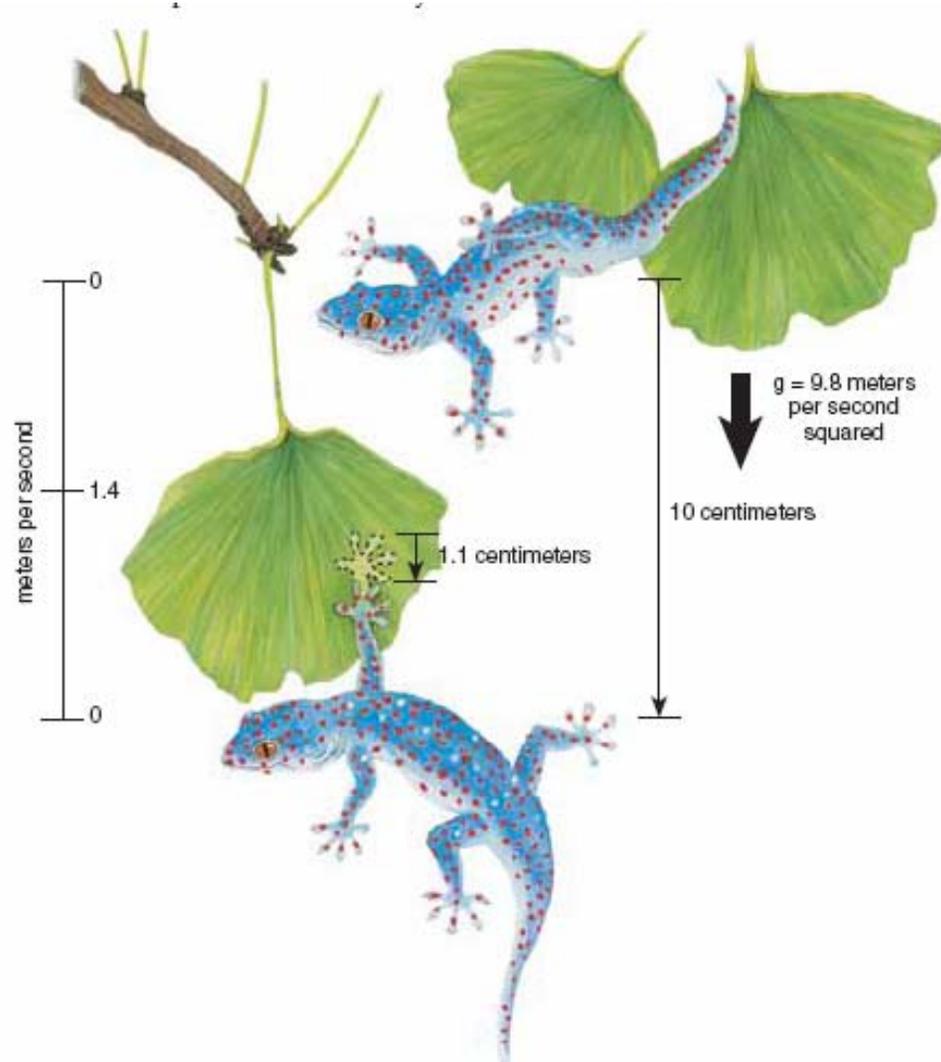
-attach and detach their toes in milliseconds to nearly every material (not Teflon!!!)

-run on vertical and inverted, rough and smooth surfaces

-gecko toes don't degrade, foul, or attach accidentally to the wrong spot → like a pressure sensitive adhesive

-they are self-cleaning and don't stick to each other

-flatten their palm down and then unroll their toes; remove without any measurable force



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HIERACHICAL STRUCTURE OF GECKO FEET (From K. Autumn, et al. *American Scientist*, 2006, 124)

Macrostructure→



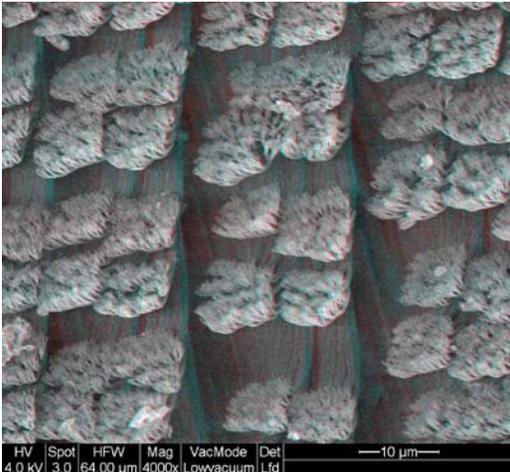
Mesostructure→



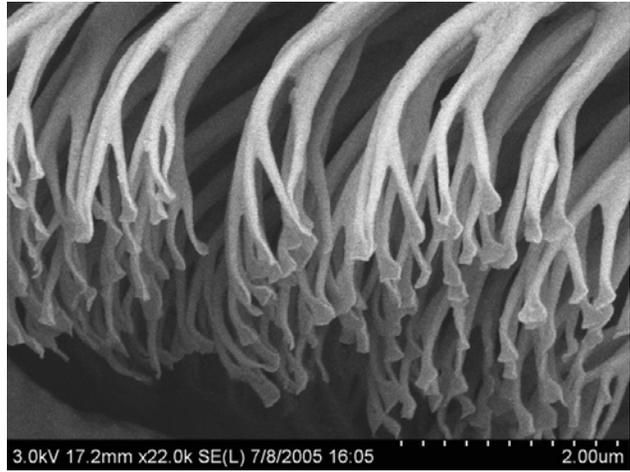
Microstructure→



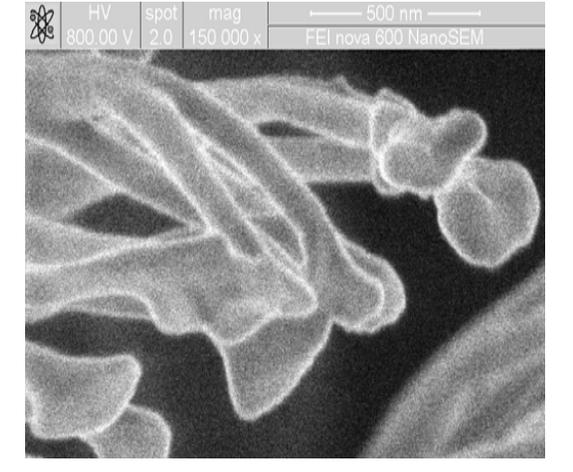
Fine microstructure→



Nanostructure→



Nanostructure→



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ADHESIVE FORCE OF A SINGLE GECKO FOOT-HAIR (From K. Autumn et al. *Nature*, **2000**, 681)

-Two front feet of a 50g Gecko can hold 2Kg, 20.1 N, 4.5 lbs

Three images from Autumn et al, *Nature* 2000, 681
removed due to copyright restrictions.

- Wanted to measure individual seta adhesion to explain macroscopic forces; couldn't get this experiment to work for months, thinking about neural control, chemicals/proteins? started applying the sequence of motions that Gecko's use (mechanical program), perpendicular **preload** and then small rearward **displacement**

- Measured force of individual seta $200 \mu\text{N}$ (can feel this) \times 6.5 million setae on all feet = 1200 N, 269 lbs, **2 medium-sized humans!!** only 3.5% of total possible adhesion needed to sustain the 2 Kg above, and $< 0.04\%$ to sustain body weight or 2000 of 6.5 million setae \rightarrow **overengineered, 3900% safety margin.**

- How do Gecko's ever take their feet off surfaces? Hair detaches automatically when **angle between setal shaft and substrate is 30 degrees** \rightarrow adhesive that is under mechanical control.

MOLECULAR ORIGINS OF ADHESION (From Autumn, et al. *PNAS* 2002 99, 19, 12252)

Theories :

× **mechanical interlocking**; nanoscale velcro hooking→ molecularly smooth Si wafers

× **suction cups**→experiments done in vacuum

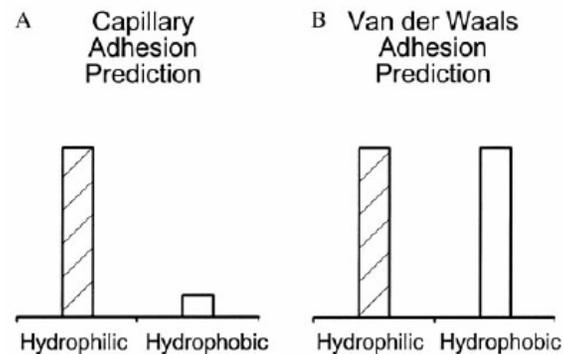
× secretion of a **protein adhesive**→lack glandular tissue in toes

× **capillarity forces** due to bridging water meniscus

-**van der Waals** forces (short range)

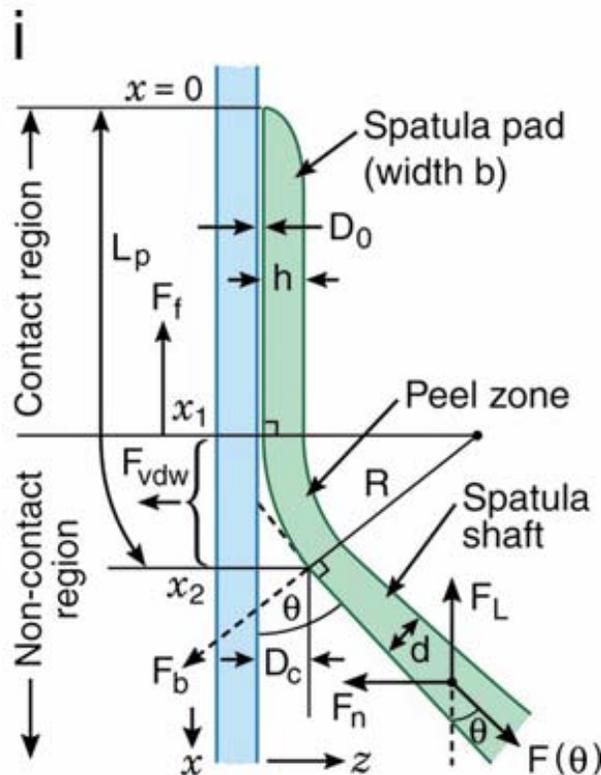
experiment hydrophilic (Si wafer) versus hydrophobic surfaces (GaAs, but is also polarizable) →Geckos stuck to both, **hence concluded VDW interactions dominate**

-**More dependent on geometry of structure rather than chemistry**



Courtesy of National Academy of Sciences, U. S. A. Used with permission. Source: Autumn, K., et al. "Evidence for van der Waals Adhesion in Gecko Setae." *PNAS* 99, no. 19 (September 17, 2002): 12252-12256. © 2002, National Academy of Sciences, U.S.A.

THEORETICAL ASPECTS OF GECKO ADHESION (From Tian, et al. *PNAS*, **2006**, 103, 51, 19320)



F_f = Friction force

F_{VDW} = van der Waals force

$F(\theta)$ = peeling force along spatula shaft

F_L = lateral component of peeling force along spatula shaft

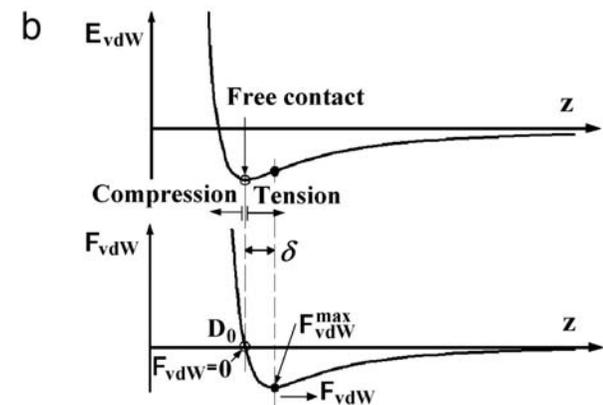
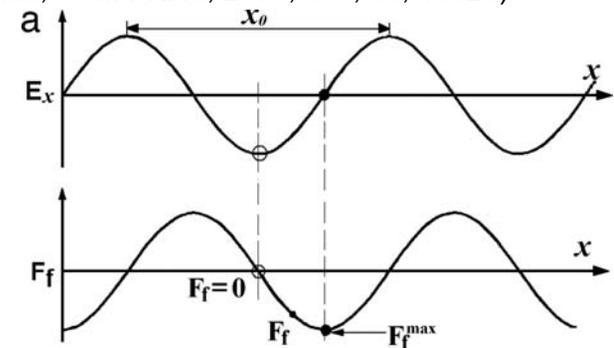
F_n = normal component of peeling force along spatula shaft

F_b = resistance to bending = negligible

(i) **contact regime**, LJ equilibrium, VDW balanced by short range atomic repulsion

(ii) transition "**peel zone**"; integrated F_{VDW} balanced by $F(\theta)$ (part of noncontact regime)

(iii) $x > x_2$ F_{VDW} negligible (part of noncontact regime), $F = F(\theta)$



$$E_x = E_0 \sin\left(\frac{2\pi x}{x_0}\right) \rightarrow F_x = F_f = \left(\frac{2\pi E_0}{x_0}\right) \cos\left(\frac{2\pi x}{x_0}\right)$$

x_0 = critical spacing related to atomic lattice, molecular or asperity dimensions on the spatula

$$P(D)_{SFC-SFC}(VDW) = \frac{-A}{6\pi D^3}$$

Courtesy of National Academy of Sciences, U. S. A. Used with permission. Source: Tian, Y., et al. "Adhesion and Friction in Gecko Toe Attachment and Detachment." *PNAS* 103, no. 51 (2006): 19320-19325. © 2002, National Academy of Sciences, U.S.A.