
Fluids - B

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Outline

- > **Review of last time**
- > **Poiseuille flow**
- > **Stokes drag on a sphere**
- > **Squeezed-film damping**
- > **Electrolytes & Electrokinetic separations**

Last time

> Surface Tension

- Force at a liquid-fluid interface

$$\tau = \eta \frac{\partial U_x}{\partial y}$$

> Viscosity

- Constitutive property relation shear stress to shear rate

$$\frac{D}{Dt} = \frac{\partial}{\partial t} + \mathbf{U} \cdot \nabla$$

> Material Derivative

- Time derivative taking into effect convection

$$\frac{D\rho_m}{Dt} + \rho_m \nabla \cdot \mathbf{U} = 0$$

> Mass continuity

> Navier-Stokes Equation

- Fundamental relation for Newtonian fluids

$$\rho_m \frac{D\mathbf{U}}{Dt} = -\nabla P + \eta \nabla^2 \mathbf{U} + \frac{\eta}{3} \nabla(\nabla \cdot \mathbf{U}) + \rho_m \mathbf{g}$$

> Reynolds Number

- The MOST IMPORTANT dimensionless number

$$Re = \frac{\rho_m LU}{\eta}$$

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Poiseuille Flow

- > Pressure-driven flow through a pipe
 - In our case, two parallel plates
- > Velocity profile is parabolic
- > This is the most common flow in microfluidics
 - Assumes that $h \ll W$

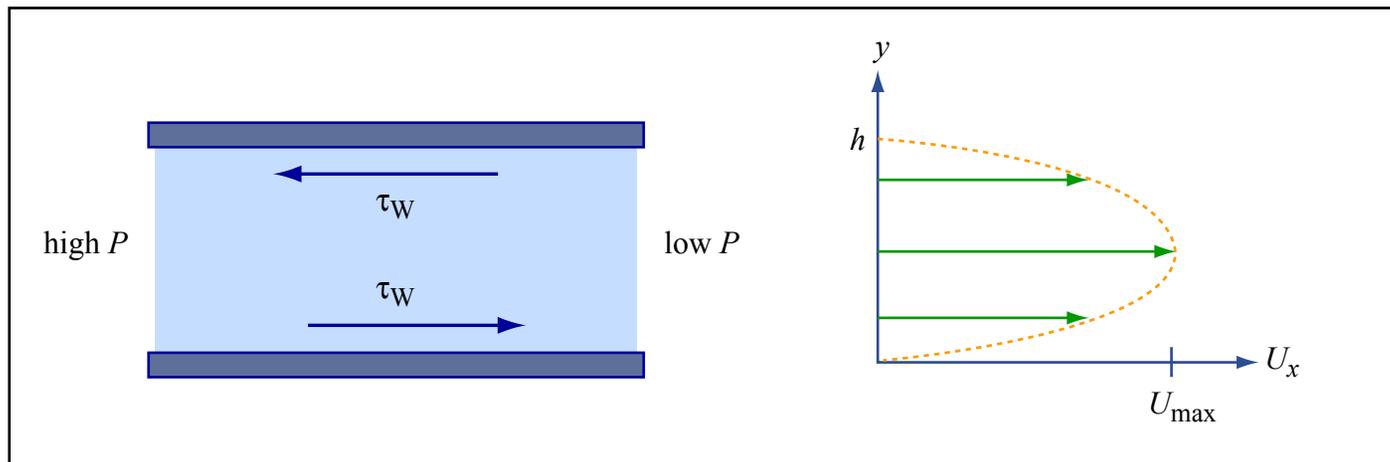


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Adapted from Figure 13.5 in Senturia, Stephen D. *Microsystem Design*. Boston, MA: Kluwer Academic Publishers, 2001, p. 329. ISBN: 9780792372462.

Solution for Poiseuille Flow

> Assume

- Incompressible
- Steady
- U_x only depends on y
- Ignore gravity

$$\rho_m \left(\frac{\partial \mathbf{U}}{\partial t} + \mathbf{U} \cdot \nabla \mathbf{U} \right) = -\nabla P + \eta \nabla^2 \mathbf{U} + \frac{\eta}{3} \nabla (\nabla \cdot \mathbf{U}) + \rho_m \mathbf{g}$$

> Assume a uniform pressure gradient along the pipe

$$\frac{dP}{dx} = -K$$

> Result is Poisson's eqn

$$\frac{\partial^2 U_x}{\partial y^2} = -\frac{K}{\eta}$$

> Boundary conditions:

- Relative velocity goes to zero at the walls
 - » no-slip boundary condition

Solution for Poiseuille Flow

- > Integrate twice to get solution
- > Maximum velocity is at center
- > Can get linear flowrate [m/s] and volumetric flowrate [m³/s]
- > Can get lumped resistor using the fluidic convention
- > Note **STRONG** dependence on h
- > This relation is more complicated when the aspect ratio is not very high...

$$U_x = \frac{1}{2\eta} [y(h-y)] K$$

$$U_{\max} = \frac{h^2}{8\eta} K$$

$$Q = W \int_0^h U_x dy = \frac{Wh^3}{12\eta} K$$

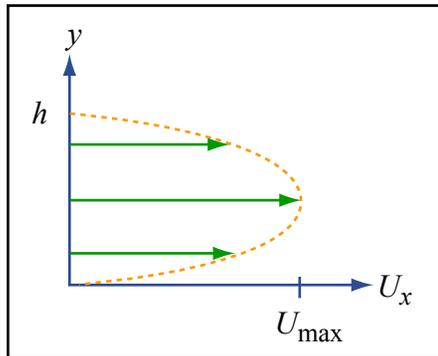


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$$\Delta P = \text{effort} = KL$$

$$\Delta P = \frac{12\eta L}{Wh^3} Q$$

$$\Rightarrow R_{Pois} = \frac{12\eta L}{Wh^3}$$

Development Length

- > It takes a certain characteristic length, called the **development length**, to establish the Poiseuille velocity profile
- > This development length corresponds to a development time for viscous stresses to diffuse from wall
- > Development length is proportional to the characteristic length scale and to the Reynolds number, both of which tend to be small in microfluidic devices

$$time \approx \frac{L^2}{\eta^*} \approx \text{Re} \frac{L}{U}$$
$$L_D \approx (time)U \approx \text{Re} L$$

A note on vorticity

- > A common statement is to say that laminar flow has no vorticity
- > What is meant is that laminar flow has no turbulence
- > Vorticity and turbulence are different
- > Can the pinwheel spin?
 - Then there is vorticity
- > Demonstrate for Poiseuille flow

$$\boldsymbol{\omega} = \nabla \times \mathbf{U}$$

Vorticity

$$U_x = \frac{1}{2\eta} [y(h-y)]K$$

$$\boldsymbol{\omega} = \mathbf{n}_y \frac{\partial U_x}{\partial z} - \mathbf{n}_z \frac{\partial U_x}{\partial y}$$

$$\boldsymbol{\omega} = -\mathbf{n}_z \frac{K}{2\eta} (h-2y)$$

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Stokes Flow

> **Steady-state flow in which inertial effects can be neglected, $Re \rightarrow 0$**

> **The result is a vector Poisson equation**

When $\rho_m \frac{DU}{Dt}$ can be neglected

> **Also called “creeping flow”**

> **Action is instantaneous**

- **No “mass” in system**
- **Incompressible: no springs**

$$\eta \nabla^2 \mathbf{U} = \nabla P^*$$

> **This is a typical approximation made in microfluidics**

Stokes' drag on a sphere

- > In creeping flow, one can solve for the flow field around a sphere placed in an initially uniform flow field
- > This can be used to find the stresses on the sphere and sum them to find the total drag
- > This is called the Stokes' drag
- > This is often the predominant particle force in microfluidic systems
- > See Deen's text for derivation

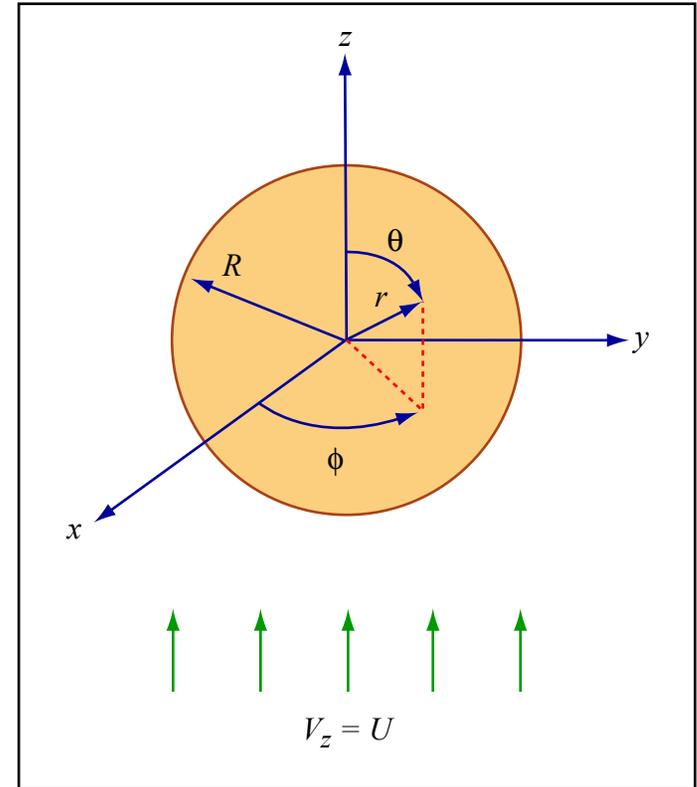
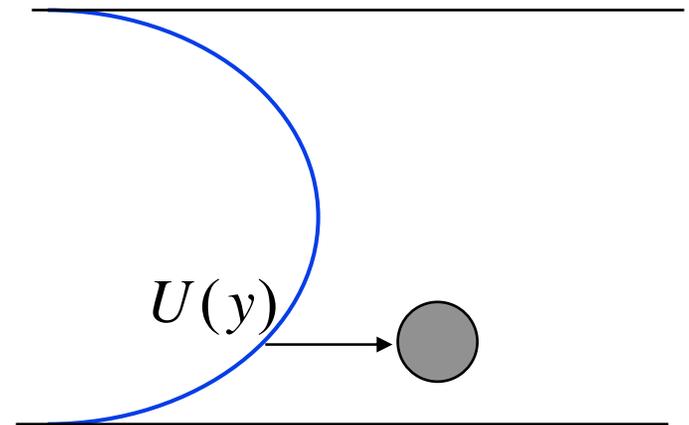


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$$F_d = 6\pi\eta R U$$

Stokes' drag on a sphere

- > This is strictly valid **ONLY** in a uniform flow
- > These are hard to make...
- > Instead, we can use $U_x(y)$ of the parabolic flow profile to calculate a height-dependent drag force
- > This approach fails when the particle is too big
- > Instead, take advantage of published solutions
 - Shear flow: Goldman et al., *Chem. Eng. Sci.* **22**, 653 (1967).
 - Poiseuille flow: Ganatos et al. *J. Fluid Mech.* **99**, 755 (1980).



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Squeezed-Film Damping

- > This is how we will get our b (or R) for the parallel-plate actuator
- > The result of motion against a fluid boundary
 - If the fluid is incompressible, there can be a large pressure rise, so large back forces result
 - If the fluid is compressible, it takes finite motion to create a pressure rise
- > In either case, the dissipation due to viscous flow provides a damping mechanism for the motion
- > This is related to “lubrication theory”

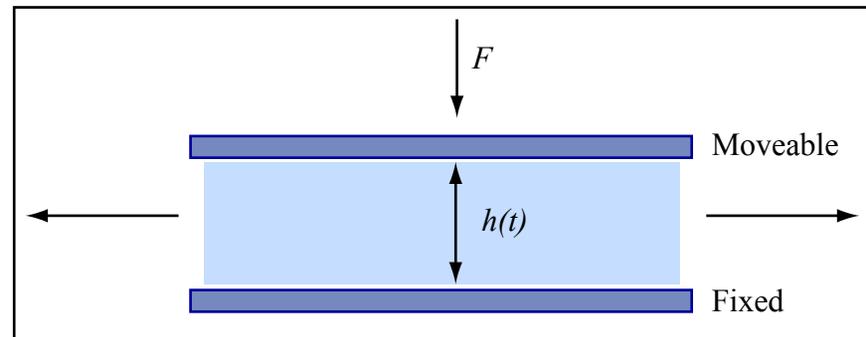


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The Reynolds Equation

> Assumptions for compressible isothermal squeezed-film damping

- One-dimensional pressure gradient: $P(r,t) = P(y,t)$ only
 - » No pressure gradient in z or along plate (x)
- Stokes flow
- Poiseuille flow profile in the plane
- Ideal gas law
- Isothermal (temperature rise due to compression is small, and heat flow to the walls is rapid)
- No-slip BC's
- Rigid plate: $h(r,t) = h(t)$ only

$$\frac{\partial(Ph)}{\partial t} = \frac{h^3}{12\eta} \left(\frac{1}{2} \nabla^2 P^2 \right)$$

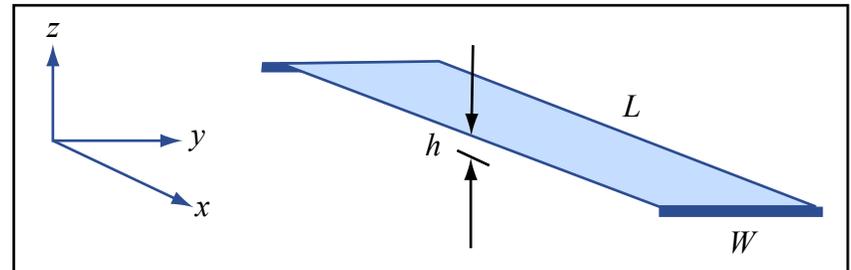


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> The result is a version of the Reynolds equation

Example: Rigid Plate Damping

- > Now, assume small motions
 - Linearize
- > The result is (guess what!) the **heat-flow equation**

$$\frac{\partial(Ph)}{\partial t} = \frac{h^3}{12\eta} \left(\frac{1}{2} \nabla^2 P^2 \right)$$

If we linearize:

$$h = h_0 + \delta h \quad P = P_0 + \delta P$$

$$\text{Normalize: } \xi = \frac{y}{W} \quad \hat{p} = \frac{\delta P}{P_0}$$

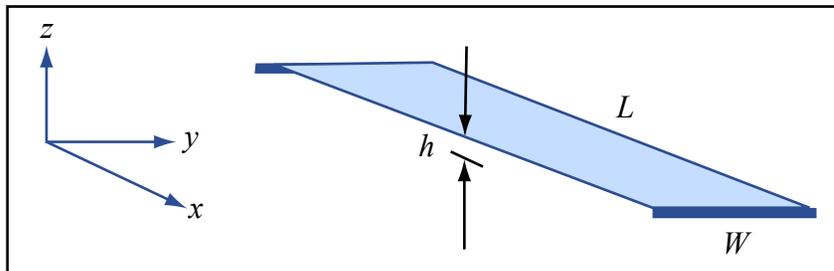


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$$\frac{\partial \hat{p}}{\partial t} = \frac{h_0^2 P_0}{12\eta W^2} \frac{\partial^2 \hat{p}}{\partial \xi^2} - \frac{\dot{h}}{h_0}$$

Suddenly Applied Motion

- > We already solved a very similar problem
 - Impulse of heat into 1-D resistor
- > Now we have a velocity impulse: sudden change of height
- > We get a series of 1st-order terms, as before
- > Only need 1st term, which is an RC circuit
- > R for viscous damping
- > C for gas compressibility
- > Details are in the book

$$F(s) = \left[\frac{96\eta LW^3}{\pi^4 h_o^3} \sum_{n \text{ odd}} \frac{1}{n^4} \frac{1}{1 + \frac{s}{\alpha_n}} \right] sz(s)$$

$$\alpha_n = \frac{n^2 \pi^2 h_o^2 P_o}{12\eta W^2} = \omega_n$$

$$b = \frac{96\eta LW^3}{\pi^4 h_o^3}$$

$$\omega_c = \frac{\pi^2 h_o^2 P_o}{12\eta W^2}$$

n=1

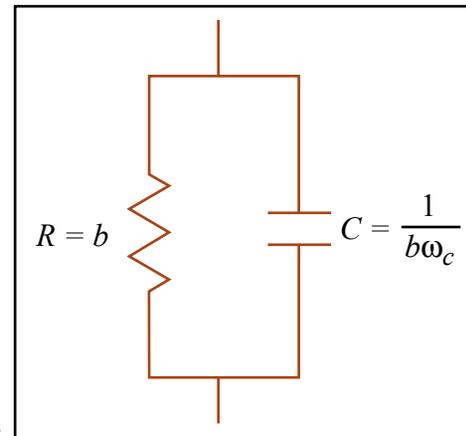


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- > **Electrolytes & Electrokinetic separations**

Electrokinetic Phenomena

- > **It is a coupled-domain problem in which electrostatic forces result in fluid flow (and vice versa)**
- > **Start with electrolytes, move into double layer, and finally show how to manipulate the double layer**
- > **It is the driver behind ALOT of early micro-TAS work**

Electrolytes

- > **Electrolytes: liquids with mobile ions**
 - **Examples: water, PBS**
- > **Ions can move via concentration gradients (diffusion) or electric fields (drift)**
- > **Macroscopically, the liquid is approximately charge-neutral (called quasineutrality)**

Ion mobility (cm²/V-s)

$$N_i = \frac{z_i}{|z_i|} u_i C_i E - D_i \nabla C_i$$

Flux (cm⁻²-s⁻¹)

Valence

Concentration (cm⁻³)

Diffusivity (cm²/s)

$$\rho_e = \sum_i z_i q_e C_i \approx 0 \text{ in the bulk}$$

In neutral regions: $\nabla^2 \phi = \frac{-\rho_e}{\epsilon} = 0$

For a binary electrolyte (e.g., NaCl in water)

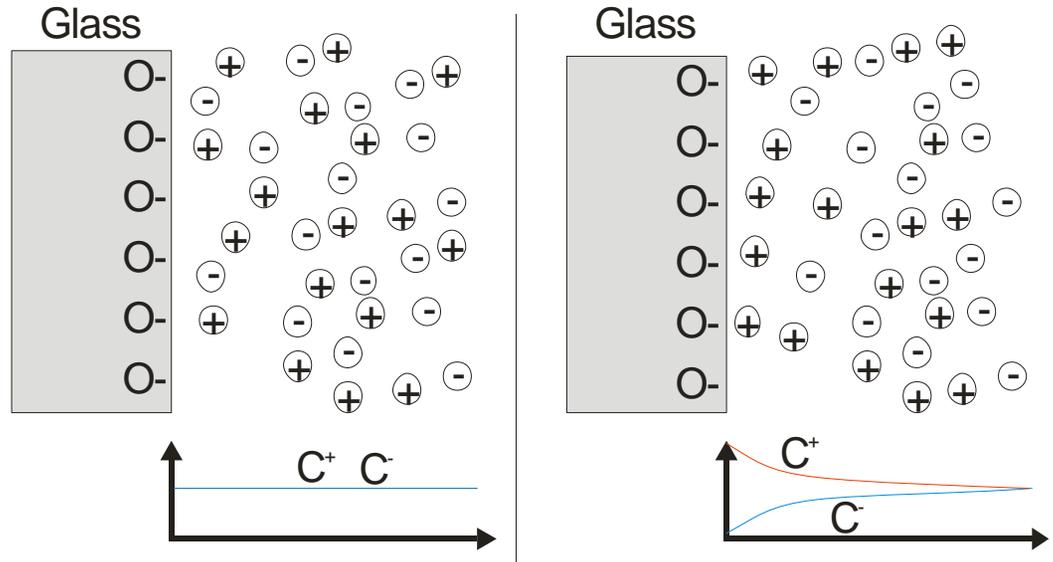
$$N_+ = u_+ C_+ E - D_+ \nabla C_+$$

$$N_- = -u_- C_- E - D_- \nabla C_-$$

$$\rho_e = q_e (C_+ - C_-) \approx 0$$

Electrolytes

- > Surfaces with fixed charge can lead to net space charge in the liquid
- > Diffusion competes with drift, and at equilibrium, Boltzmann distribution follows
- > This leads to the Poisson-Boltzmann equation



$$N_i = \frac{z_i}{|z_i|} u_i C_i E - D_i \nabla C_i = 0 \quad \text{at equilibrium}$$

$$\Rightarrow C_i(x) = C_{i,o} e^{\frac{z_i q_e (\phi(x) - \phi_o)}{k_B T}}$$

Near the wall: $\nabla^2 \phi = -\frac{\rho_e}{\epsilon} \Rightarrow \nabla^2 \phi = -\frac{1}{\epsilon} \sum_i z_i q_e C_{i,o} e^{\frac{z_i q_e (\phi - \phi_o)}{k_B T}}$

Diffuse Double Layer

Express in terms of $\hat{\phi} = \phi - \phi_0$:
$$\nabla^2 \hat{\phi} = -\frac{1}{\epsilon} \sum_i z_i q_e C_{i,o} e^{-\frac{z_i q_e \hat{\phi}}{k_B T}}$$

Expand for small potential variations

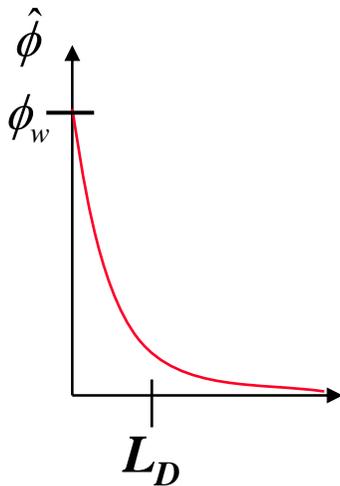
$$\nabla^2 \hat{\phi} \cong -\frac{1}{\epsilon} \sum_i z_i q_e C_{i,o} + \frac{q_e^2}{\epsilon k_B T} \left[\sum_i z_i^2 C_{i,o} \right] \hat{\phi}$$

Assuming the reference region is charge neutral

$$\nabla^2 \hat{\phi} = \frac{1}{L_D^2} \hat{\phi}$$

$$\hat{\phi} = \phi_w e^{\pm z/L_D}$$

For binary monovalent electrolyte (e.g., NaCl)



Debye Length

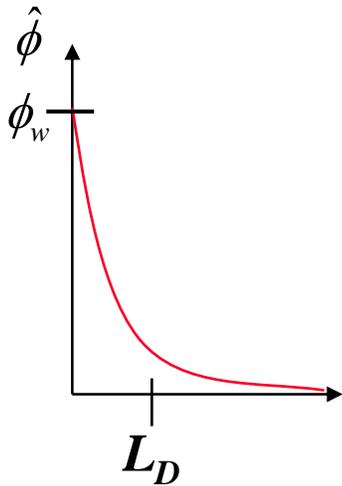
$$\frac{1}{L_D} = \sqrt{\frac{q_e^2}{\epsilon k_B T} \left[\sum_i z_i^2 C_{i,o} \right]} = \sqrt{\frac{2q_e^2 C_{i,o}}{\epsilon k_B T}}$$

$$L_D \sim \frac{1}{\sqrt{C_{i,o}}}$$

$L_D = 1$ nm for
0.1 M NaCl

Double-layer charge

- > We can calculate the total charge in the double layer
- > It must balance the charge at the wall



Near the wall: $\hat{\phi} = \phi_w e^{-\frac{z}{L_D}}$

The charge density is

$$\nabla^2 \hat{\phi} = -\frac{\rho_e}{\epsilon} \Rightarrow \rho_e = -\frac{\epsilon}{L_D^2} \hat{\phi}$$

Total charge per unit area
in diffuse layer

$$\sigma_d = \int_0^{\infty} \rho_e dz = -\frac{\epsilon \phi_w}{L_D} = -\sigma_w \Rightarrow \hat{\phi} = \frac{\sigma_w L_D}{\epsilon} e^{-\frac{z}{L_D}}$$

DL charge/area
[C/cm²]

wall charge/area
[C/cm²]

Actual Double Layers

- > The actual situation is MUCH more complicated
- > Some ions are tightly or specifically adsorbed, forming the Stern layer and screening the wall charge
- > The rest distribute in a diffuse double-layer: the Gouy-Chapman layer
- > This is an active area of research

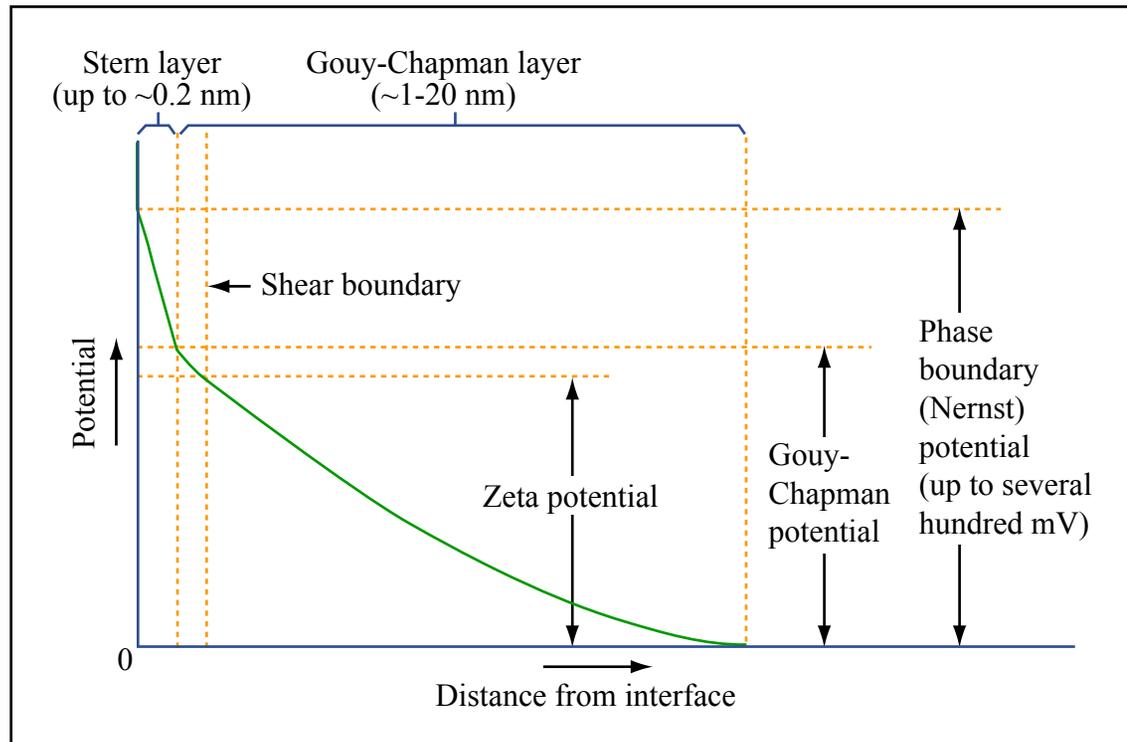


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Electroosmotic Flow

- > An axial electric field exerts a force on the charge in the diffuse double layer, which drags the fluid down the pipe

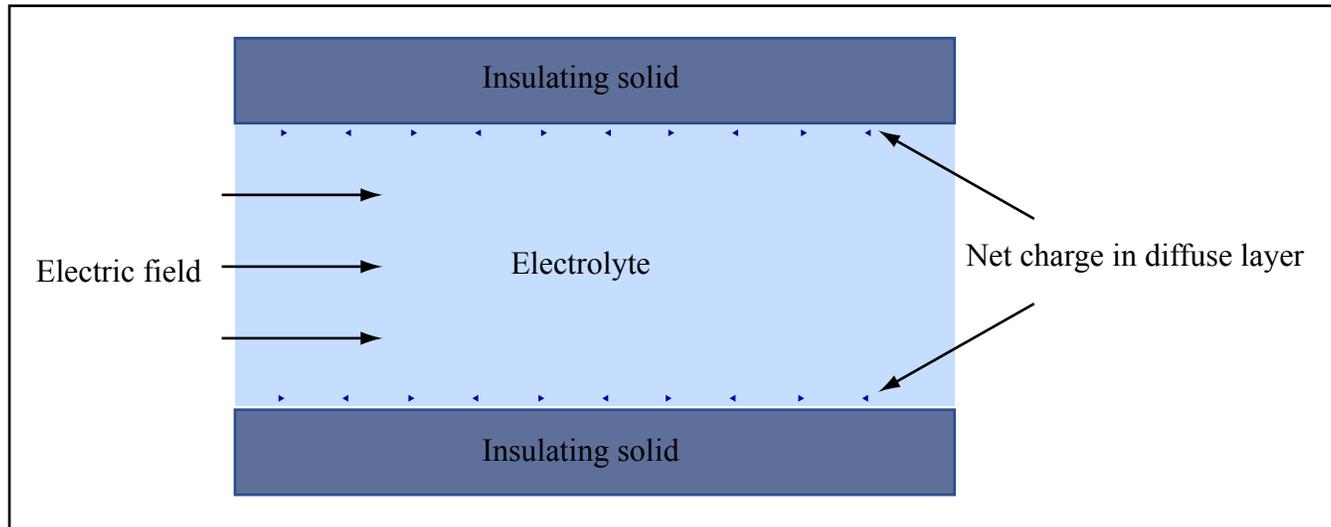


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Analysis of Electroosmotic Flow

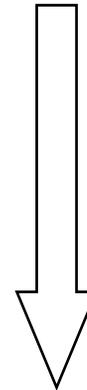
> Assume

- Creeping flow
- One dimensional flow $U_x(z)$
- No pressure drop
- Electrical body force

> Express charge density in terms of wall surface charge density

> Set up differential equation

$$\eta \nabla^2 U_x = \rho_e E_x$$



$$\begin{aligned} \rho_e &= -\frac{\varepsilon}{L_D^2} \hat{\phi} \\ &= -\frac{\sigma_w}{L_D} e^{-\frac{z}{L_D}} \end{aligned}$$

$$\frac{d^2 U_x}{dz^2} = \left(\frac{\sigma_w}{\eta L_D} e^{-\frac{z}{L_D}} \right) E_x$$

Analysis of Electroosmotic Flow

> This leads to a PLUG-FLOW profile

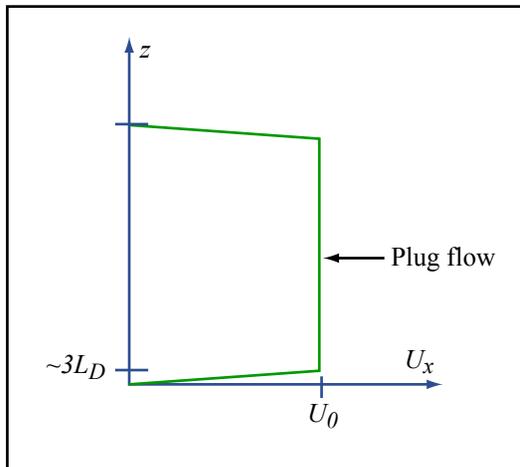
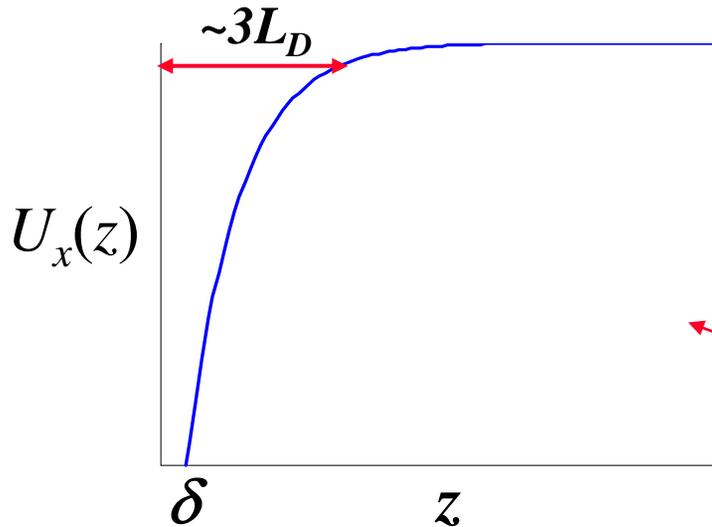


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$$\frac{d^2 U_x}{dz^2} = \left(\frac{\sigma_w}{\eta L_D} e^{-\frac{z}{L_D}} \right) E_x$$

Integrate twice, and use boundary conditions:

$$U_x(z = \delta) = 0 \quad \hat{\phi}(\delta) = \zeta \quad \left. \frac{dU_x}{dz} \right|_{h/2} = 0$$

$$U_x = -\frac{\sigma_w L_D}{\eta} \left(e^{-\frac{\delta}{L_D}} - e^{-\frac{z}{L_D}} \right) E_x$$

$$\hat{\phi}(z) = \frac{\sigma_w L_D}{\varepsilon} e^{-\frac{z}{L_D}}$$

$$U_x = -\frac{\varepsilon}{\eta} (\zeta - \hat{\phi}) E_x$$

Analysis of Electroosmotic Flow

> This leads to a PLUG-FLOW profile

> Flow depends on **zeta potential** ζ

- The potential at the slip plane δ
- Which is in a different place than the wall, the Stern layer, or L_D
- It is what is measured experimentally
- For $h \gg L_D$, one typically assumes

$$\delta = 0$$

$$\phi_w = \zeta$$

$$U_x = -\frac{\varepsilon}{\eta} (\zeta - \hat{\phi}) E_x$$

$$U_0 = -\frac{\sigma_w L_D}{\eta} e^{-\frac{\delta}{L_D}} E_x = -\frac{\varepsilon \zeta}{\eta} E_x$$

for $z > 3L_D$

Electrophoresis

- > This is just electroosmosis around a solid surface
- > Each ionic species has its own mobility
- > Therefore, in an electrolyte in which there is a net electric field, ions will drift at various rates
- > This is the basis of a separation technology called **electrophoretic separation**

$$U_{ep} = \left(\frac{\varepsilon \zeta}{\eta} \right) E_x = \mu_{ep} E_x$$

For a “large” particle

More generally

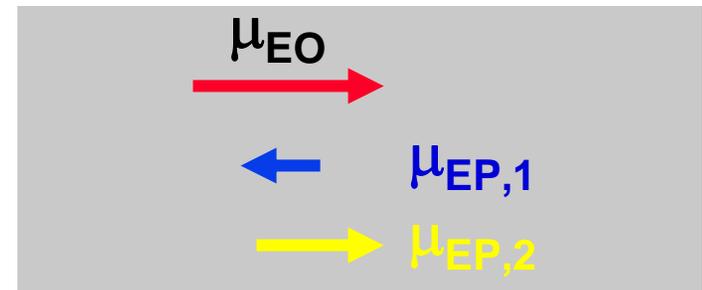
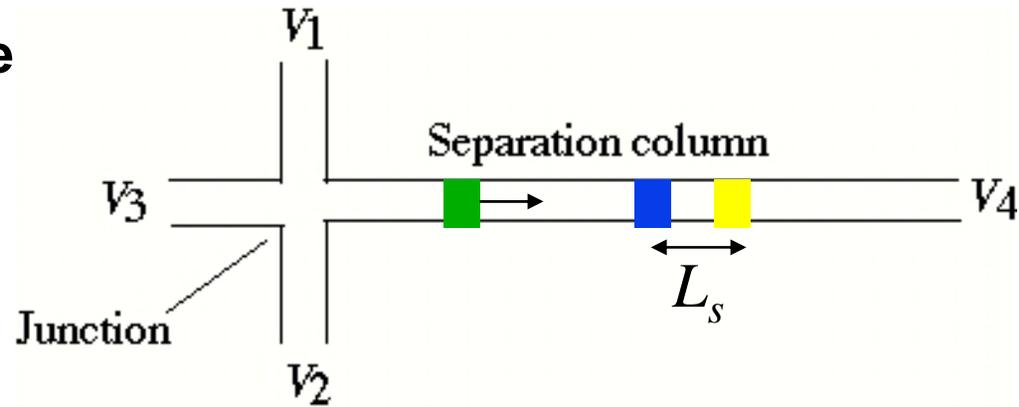
Electrokinetic separation

- > This was **THE** original driver for micro-TAS
 - TAS = total analysis systems
- > Create fully integrated microsystems that would go from “sample to answer”
- > The **KEY** enabler was the integration of non-mechanical valves with the separation column
 - This creates extremely narrow sample plugs
- > Also important is the ability to multiplex
- > The *actual* sample prep was (and is) usually ignored
- > This is/was the *raison d'être* of Caliper & Aclara
- > Aclara died, unclear if Caliper will succeed

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Electrokinetic separation

- > We use a channel-crossing structure to select a sample plug
- > Then we switch voltages to drag the plug down a separation column
- > In one approach, EO and EP co-exist, but EO dominates
- > Different species travel at different rates → separation



$$L_s = t_{sep} \left| \mu_{EP,1} - \mu_{EP,2} \right| E_x$$

↑
Separation time

Schematic Illustration

- > The key discovery was that liquid samples could be controlled with voltages
- > This allows one to valve and pump liquids
 - Create small sample plugs

ORNL movie

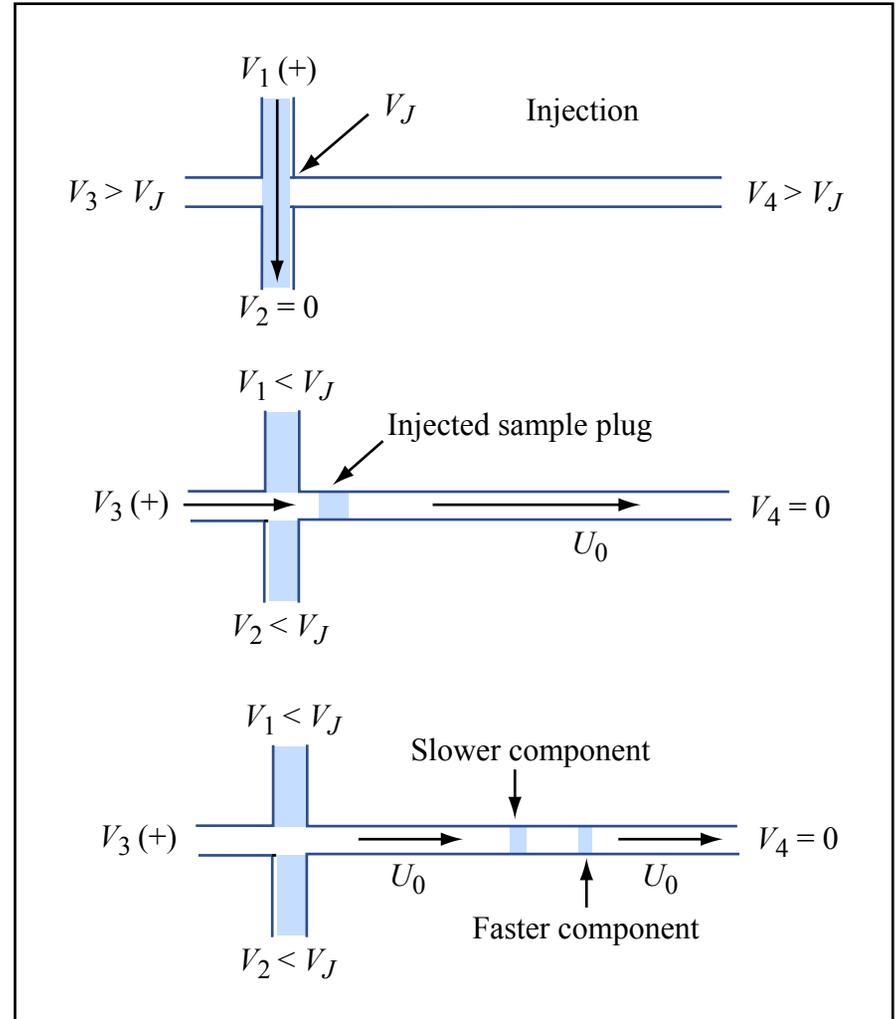
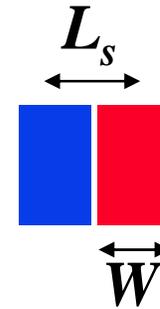


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Electrokinetic separation

- > The “macro” technology is “conventional” capillary electrophoresis
 - Most notably used in the human genome project
- > Sample loading is the problem
- > To separate two species (ignoring diffusion), we need
- > Smaller starting W means we can use a shorter channel
- > And get a faster separation

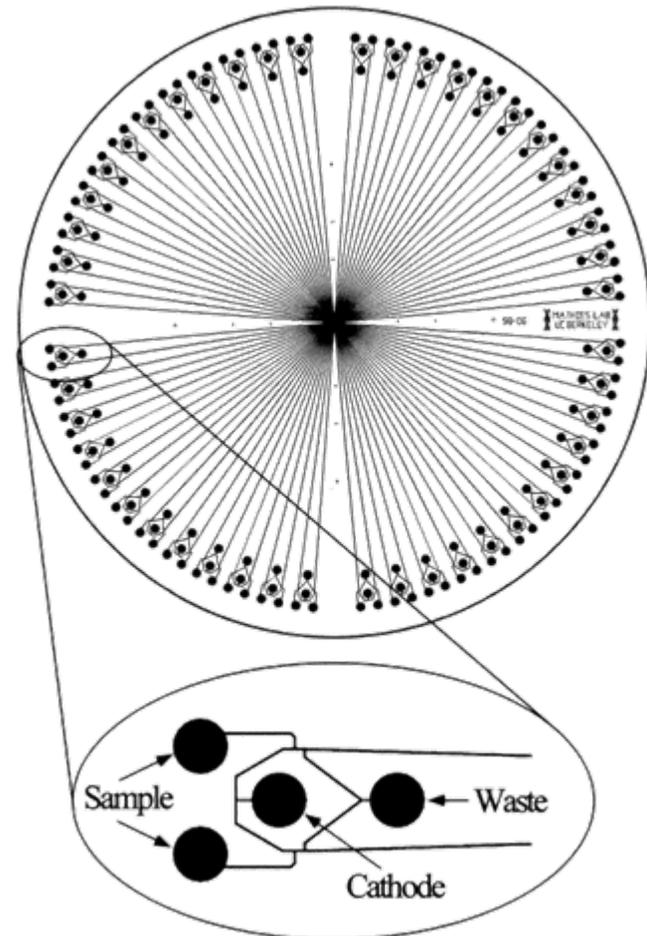
$$L_s \geq W$$



Electrokinetic separation

- > **Conventional capillaries have larger W because injection is not integrated**
 - Though this is always getting better
- > **This thus requires a longer channel**
- > **Microfab also allows for integration**

Richard Mathies (UCB)



Courtesy of Richard A. Mathies. Used with permission.

An aside on electrodes

- > **Current in the electrolyte is carried by ions**
- > **Current in the wire is carried by electrons**
- > **At the surface, something must happen to transfer this current**
- > **This is electrochemistry and the typical byproduct are gases → bubbles**

Electrokinetics

> That is just the beginning

> Dielectrophoresis

- Force on dipoles in non-uniform electric fields
- Can use AC fields
- Can hold things in place

> Other phenomena

- Electrohydrodynamics
- Electrowetting
- Induced-charge electrophoresis/electroosmosis

Comparing EOF & pressure-driven flow

- > If you are designing a chip that needs to move liquids around, which method is best?
- > Issues to consider
 - Flowrate scaling
 - Liquid composition
 - System partitioning
 - Materials
 - Species Transport

Comparing EOF & pressure-driven flow

> Flowrate

- Water in rectangular SiO₂ channel
- h varies, W=1000 μm, L=2 cm, ε=80ε₀, ζ=50 mV
- Drive with E_x=100 V/cm, ΔP=5 psi

$$U_0 = -\frac{\varepsilon\zeta}{\eta} E_x$$

$$Q = U_0 h W = -\frac{\varepsilon\zeta h W}{\eta} E_x$$

$$Q = \frac{Wh^3}{12\eta L} \Delta P$$

$$U_0 = \frac{Q}{A} = \frac{h^2}{12\eta L} \Delta P$$

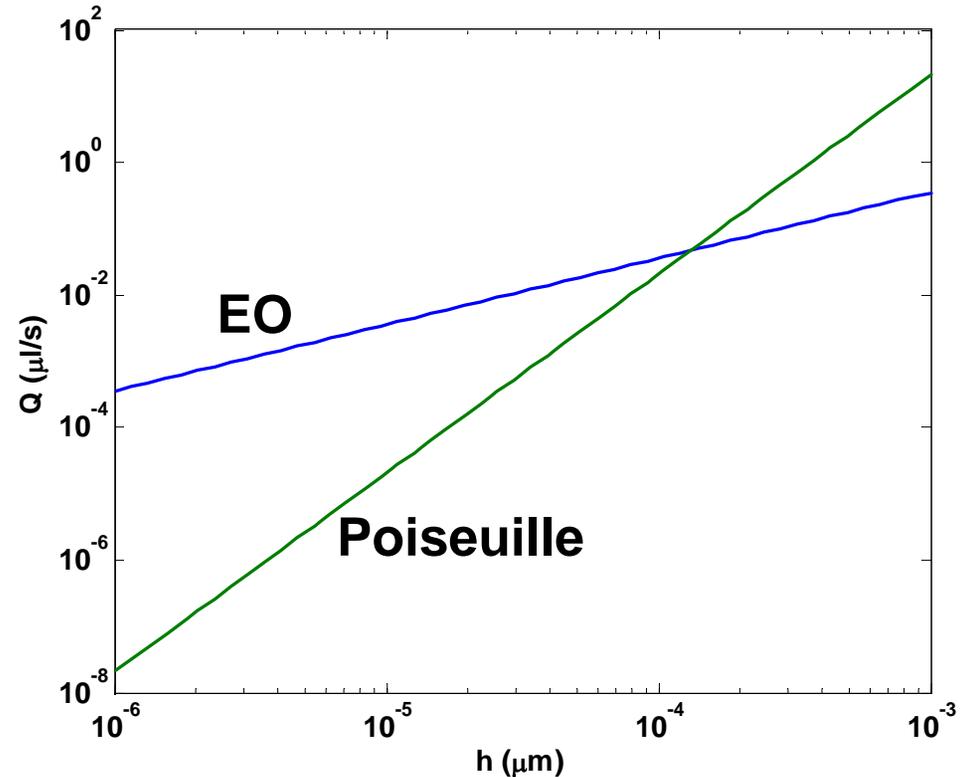
Comparing EOF & pressure-driven flow

> Scaling

- Pressure-driven flow larger in large channels
 - » Due to cubic dependence of flow resistance
- EO flow larger in small channels

> Both scale equivalently with channel length

- Larger L requires more voltage and higher pressure to get same flow



Comparing EOF & pressure-driven flow

- > Other issues typically matter more
- > Valving
 - EOF has “built-in” valving using E-fields
 - Poiseuille needs mechanical valves
- > Liquid limitations
 - Poiseuille flow can pump any liquid
 - EOF ionic strength limits
 - » Debye length depends on $1/\sqrt{C_0}$
 - » Increasing ionic strength decreases L_D and thus EOF
 - » Typically use ~10-100 mM salt buffer
 - EOF pH limits
 - » pH affects wall charge → affects EOF
 - » Typically use pH ~7 buffers

Comparing EOF & pressure-driven flow

> Materials issues

- Poiseuille flow can use any material
- EOF requires defined (and stable) surface charge
 - » Best is silica (or at least glass)
 - » Polymers are more difficult
 - » Surface charge can change as molecules adsorb, etc.

> System partitioning

- Both approaches involve off-chip components
 - » Electrodes for EOF
 - » Pumps for Poiseuille flow
- Integrating complete lab-on-a-chip usually BAD idea
 - » EOF: Easier to use external electronics
 - » Poiseuille: Pumps hard to make on-chip

Comparing EOF & pressure-driven flow

> Transport limitations

- EOF
 - » Will separate molecules as they are convected downstream
 - » Species in flow must tolerate E-fields (DNA/proteins OK, cells not so good)
 - » Plugs remain plugs
- Poiseuille flow
 - » Will not separate molecules
 - » Objects in flow must tolerate shear/pressure (DNA/proteins OK, cells OK depending on shear/pressure)
 - » Will distort plugs

What's next

- > **We will discuss the behavior of the stuff in the liquids**
- > **How to manipulate that stuff at the microscale**
- > **And then we head into system-level issues**
 - **Feedback, Noise, etc.**