

Chemolithotrophic Organisms (cont.)

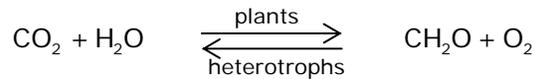
Exist @ interface of aerobic & anaerobic organisms

- Sulfur-oxidizers: $H_2S, S^0, S_2O_3^{2-} \rightarrow SO_4^{2-}$ / O_2, NO_3^-
- Iron-oxidizers: $Fe(II) \rightarrow Fe(III)$ / O_2, NO_3^-
 ↳ Only efficient @ acidic pH
- Methane-oxidizers: $CH_4 \rightarrow CO_2$ / O_2
 e⁻ acceptors

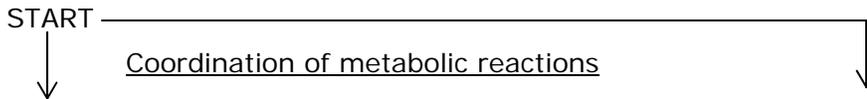
(Methanotrophs) obligate methane oxidizers

Biogeochemical Cycles

e⁻ tower = model of the biosphere



Carbon dominates all other biogeochemical cycles



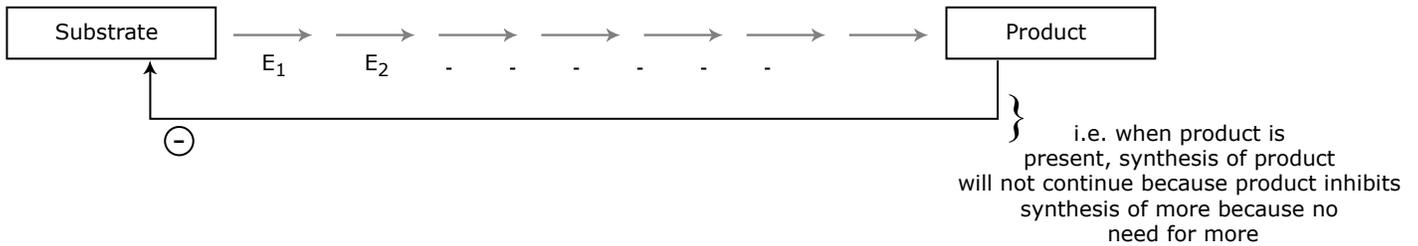
Regulation:

1. Enzyme activity
 2. Enzyme level
 3. Global control networks
1. Enzyme activity: controlled how?
 - o Specific proteases destroy enzymes (destruction)
 - o Activation: cleavage of inactive precursor
 - o Modification:
 - Covalent modification (example phosphorylation)
 - Reversible modification

↳ Effector molecule (product) must bind possibly in addition to substrate.

Allosteric enzymes → enzyme inhibition

↳ Modified by feedback inhibition



→ Typically at 1st enzyme in pathway

Control of central metabolism

- Intermediates are allosteric effectors
 - ↳ Example: phosphoenolpyruvate is allosteric effector for phosphofructokinase
- ATP, NADH, NADPH – allosteric effectors because amounts present indicate cell energy status

Cells energy status:

$$\text{energy charge (EC)} = \frac{[\text{ATP}] + \frac{[\text{ADP}]}{2}}{[\text{ATP}] + [\text{ADP}] + [\text{AMP}]}$$

} in cells, ~ 0.87-0.95

Generalization: if energy charge is high, cell will divert pathway towards more biosynthesis. If energy charge is low, cell will divert pathway towards fueling.

$$\text{Catabolic Reduction Charge} = \frac{[\text{NADH}]}{[\text{NADH}] + [\text{NAD}^+]} \sim \underbrace{0.3-0.7}_{\text{range}}$$

$$\text{Anabolic Reduction Charge} = \frac{[\text{NADPH}]}{[\text{NADPH}] + [\text{NADP}^+]} \sim 10 \text{ fold higher than CRC}$$