## Practice Exam #4 5.13: Organic Chemistry II Fall 2003

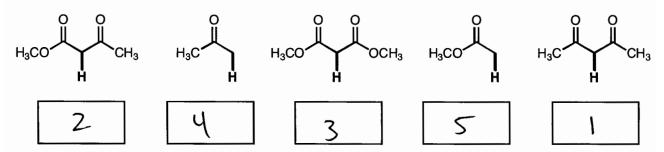


 Midterm exam #4 will be held on Friday, December 5, from 12–1pm.

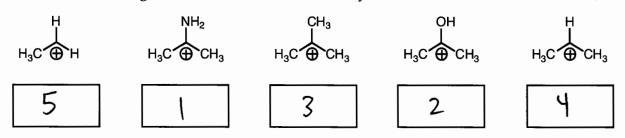
- Ten points will be subtracted from your exam if you take it in the wrong room.
- Notes, books, and calculators will not be allowed in the exam.
- You will be free to use molecular models during the exam.
- You will be given a periodic table.
- The exam will cover reading (Ch. 22) and lecture material through Unit VII (Monday, December 1st). Knowledge of prior material (5.12 & 5.13) is also assumed.
- For best results, take this test as if it were your exam (without a key in 50 minutes!). That way, you'll know what areas to work on before you get to the real exam.

Practice, Practice, Practice!!!

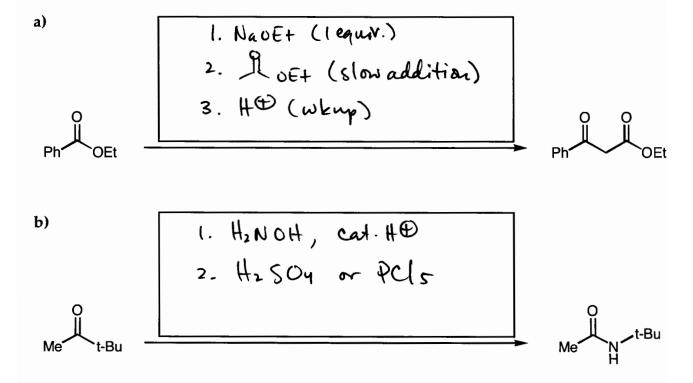
## **1. a)** Rank the following carbonyl compounds based on the acidity of the indicated $\alpha$ -proton. (1 = most acidic, 5 = least acidic)



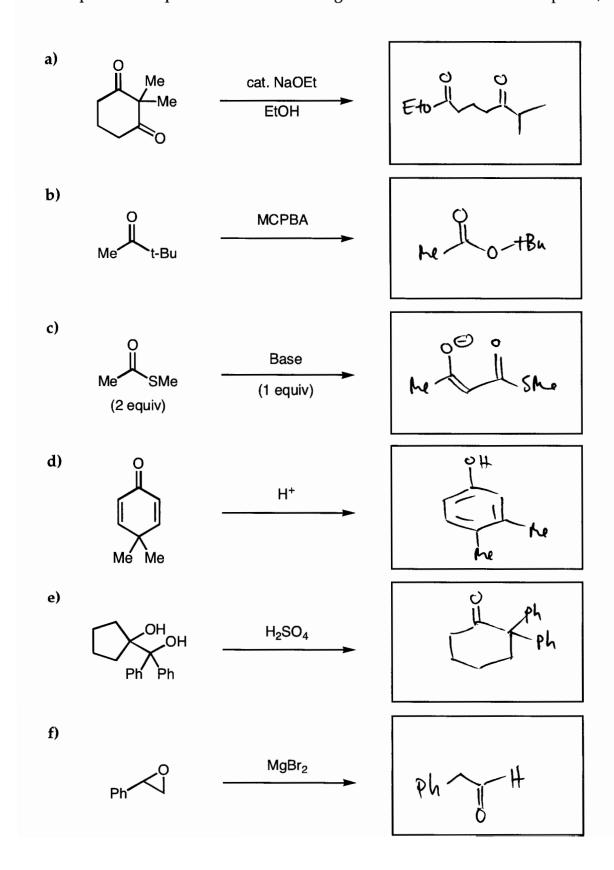
**b)** Rank the following carbocations based on stability. (1 = most stable, 5 = least stable)



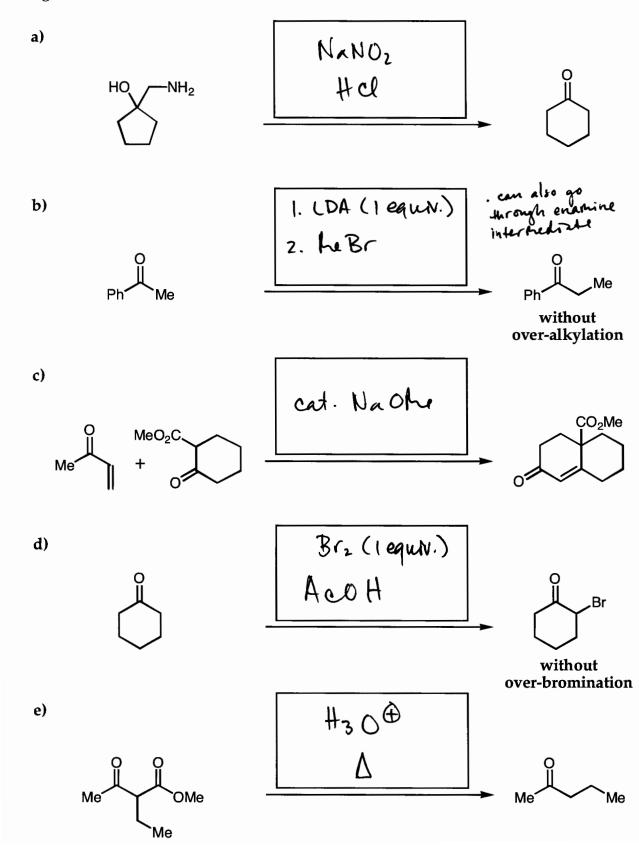
**2.** In the boxes, please provide the reagents for the illustrated transformations. Include work-up steps and be specific about quantities of reagents where relevant. More than one step may be required.

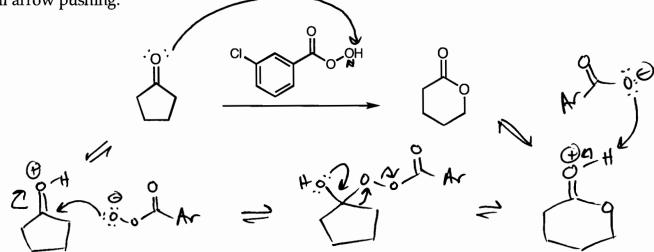


**3.** Please provide the products of the following reactions. If no reaction is expected, write "NR".



**4.** Please provide the reagents for the following transformations. Be specific about quantities of reagents where relevant.





**9.** Please provide a synthesis of the indicated compound. All of the carbon atoms should be derived from methyl acetate.

**10.** Please provide a synthesis of the indicated compound. All of the carbon atoms should be derived from isopropanol.

