Lecture #10

Lecture 10 2/25/04
References Voet& Voet, Biochemistry, p680-704 Science 232, 34-47 (1987) -Review Science 191, 150-4 (1976) -Methods
From PNAS 97 , 14188 (2000) Fidelity in PKS
Biosynthesis of Yersiniabactin (mixed PKS, NRPS) AcetylCoA (AcoA) is present in high concentration in cell- 1)Can you misload ACoA? Yes- in vitro 2) Can it be processed by hydrolysis a requirement for the normal reaction to proceed?
Made mutants of HMWP1 (one of he genes encoding the machinery to make Yersiniabactin) to show that the entire assembly line of domains is involved in hydrolysis of ACoA
Hypothesis: AcetylCoA is hydrolyzed rapidly by a "bucket brigade," using the entire protein: HMWP1
HMWP1= KS AT MeT KR ACP C MeT ₃ PCP ₃ TE (3H OH SH CH) Results
Construct AcetylCoA hydrol rate His6-HMWP1 100% 140min ⁻¹ (rate of hydrolysis 100x faster than of ACoA in solution w/out HMWP1)
His6- PCP ₃ - TE .24% -background level of hydrolysis
His6- HMWP1 – TE- CH ₃ (mutant Ser->Ala) .33%
His6- HMWP1 –(AT mutant Ser-Ala) 2.5%
His6- HMWP1-(ACP-CH ₃ mutant Ser-Ala) 7.5%

His6- HMWP1-(PCP₃-CH₃ mutant Ser-Ala) 4.4%

Conclusions:

PCP3 cannot pick up and hydrolyze ACoA HMWP1 with TE mutant Ser-Ala = no hydrolysis, so the TE domain is necessary AT,ACP and PCP domains are all necessary

Supports hypothesis

This is an intriguing result that makes you think about the question of fidelity- how do these machines achieve fidelity?

Cholesterol Biosynthesis

I will abbreviate cholesterol as "Ch"

Brown and Goldstein won the nobel prize for their seminal research in the field

Cholesterol homeostasis is key- Ch is both essential and deadly

3 Nobel Prizes have been awarded for research about Ch

1927- Wieland (structure)

1964- Bloch, Conforth, Popiack (biosynthesis)

1984- Brown and Goldstein (LDL-receptor)

Will Brown and Goldstein win another Nobel Prize? How do you sense an insoluble metabolite?- New area of research

Ch is a rigid small molecule, hydrophobic, solubility 5 microM

Functions:

- -essential constituent of membranes
- -affects physical properties of membranes
- -precursor to bile acids and steroid hormones (see handout 2d for structures of these)

When Ch homeostasis is not controlled it can be deadly

- -deposited in arteries
- -cardiovascular disease, strokes

Understanding Ch homeostasis

I Biosynthesis

II Regulation

a)LDL-receptor, receptor mediated endocytosis

b)work in progress – sterol responsive element-binding protein (SRE-BPs)

"sensor of insoluble metabolites"

Overview of stages of Ch Biosynthesis See page 1 handout 2d

EARLY

- 1) 3ACoA-> HMG-CoA
- 2) HMGCoA reductase is major regulatory step (target of statin drugs like lovastatin), takes HMG-CoA to mevalonate (C₆), uses 2NADPH
- 3) Mevalonate -> isopentenyl pyrophosphate (IPP) (C₅), decarboxylative elimination, uses 3ATP

MIDDLE

- 1) IPP is a major building block of natural products made into rubber by plants (huge oligamer with 10⁵ IPP)
- 2) isomerizes to Dimethylallyl pyrophosphate (DAPP) (used by some tRNAs)
- 3) IPP and DAPP are converted to Farnesyl –PP (C₁₅)

FarnesylPP is used in posttranslational modification of certain Ras proteins, dolichol (plays a central role in posttranslational modification by glycoslyation), CoQ quinones

4) Squalene synthase converts FarnesylPP to Squalene (C₃₀) (major regulated step, drug target)

LATE

- 1) Squalene is converted into Cholesterol
- 2) Ch is used to make bile acids, steroid hormones, vitamin D

- OH /

/ V 0PP

OPP

Structure Mevalonate

Isopentenvl PP

Dimethylallyl PP

Formation of mevalonate (page 2 handout 2d)

- 1)AcCoA reversible is converted to acetoacetylCoA via a thiolase
- 2)HMGCoA synthase converts acetoacetylCoA + ACoA to HMGCoA
- 3)HMGCoA reductase (2NADPH for hydride transfer) converts HMGCoA to mevalonate

HMGCoA reductase is the rate determining step in Ch biosynthesis! Target of lovastatinpotent reversible inhibitor of HMGCoA reductase

See structure of lovastatin page 3 of handout 2d

Structure of lovastatin mimics mevalonate- targets active site by looking like the natural product

-bottom of lovastatin is a greasy hydrophobic mess-binds tightly into a binding pocket on the enzyme (common strategy in drug design)

Page 2 of handout 2d shows the conversion of mevalonate to IPP and DPP

- -uses 3 ATP
- -2 phosphorylations of the terminal primary OH of mevalonate
- -1 phosphorylation secondary hydroxyl at C₃- makes a better leaving group
- -decarboxylation to IPP (C6 is converted to a C5 unit)