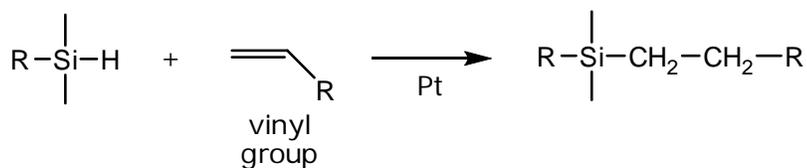
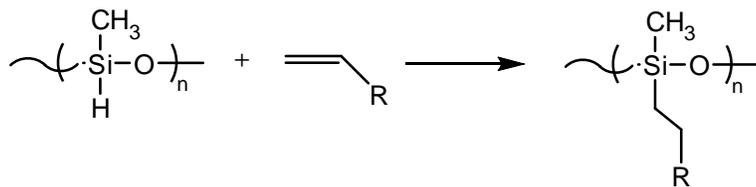


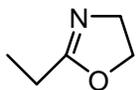
- hydrosilation
 general reaction



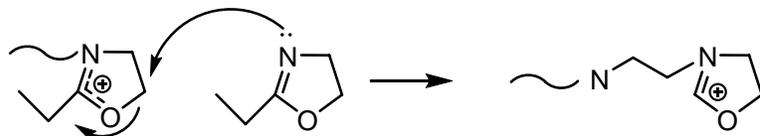
on polymers



Handout

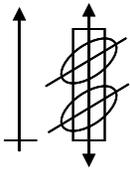


2-ethyl-2-oxazoline
 cationic ring-opening polymerization



Backbone can be converted to polyethylene amine.

Liquid Crystals (LC's)

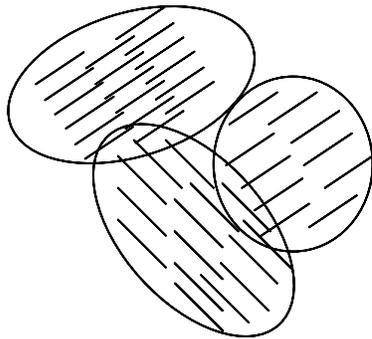


→ Board-shaped
"calamitic"

can orient preferentially



→ disc-shaped
"discotic"



crystals orientation
domains of different orientation
guide light along domains

polydomains

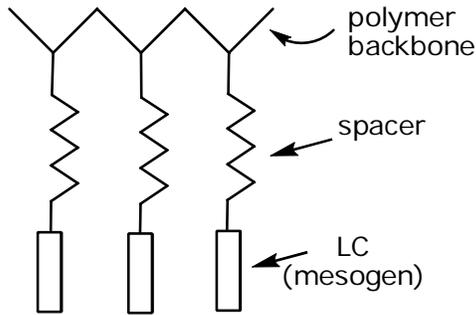
Align LC → using EM field
can allow light to pass through

Applications: displays – need mechanical stability
storage – freeze in orientation and read later
glassy polymer

→ How to attach LC to polymers

1. How to form polymer w/LC side group
2. Block copolymer: LC side chain block and glassy block

Design "Case Study" – LC Side Chain Polymers



Requirements

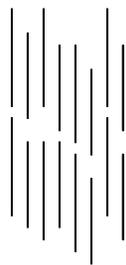
- Backbone: determines T_g
 - mechanical properties
 - chemical stability
 - can affect switching times/ response times

-Alkyl spacer group

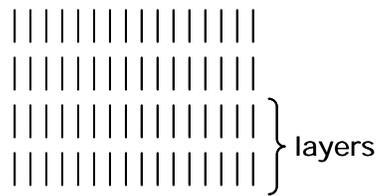
- choose spacer group: alkyl and ethylene oxide (most common)

↑ shorter
↓ leads to nematic phases

↑ more flexible
b/c CH_2O bond



Unidirectional
Ordered
⇒ nematic



Stacked: 2D order
⇒ smectic
more ordered system

Longer spacers form smectic phases
alkyl group will order

-Mesogen

- LC properties
 - optoelectronic response
 - alignment
 - time
 - dipole

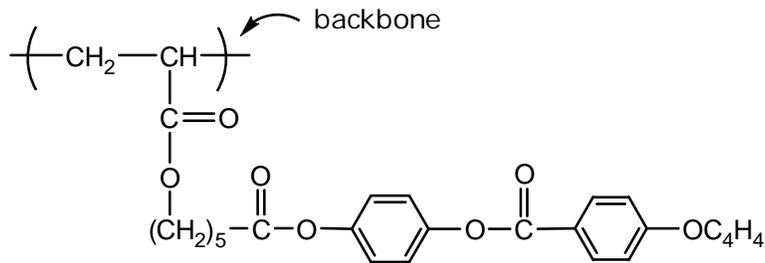
-Other factors

- does PDI matter?

For this application, it depends which phase you're going for

For LC polymers, can lose some less stable phases w/high PDI

Example:



PDI

a 1.2

b 2.9

Increasing T →

$S_F \rightarrow RN \rightarrow S_A \rightarrow N \rightarrow I$

$S_F \rightarrow N \rightarrow I$

close for
both PDI

S_F = smectic F

I = isotropic clearing point

N = nematic

- High MW/Low MW?
Oligometric → stable image
High MW → storage
- Selected polymer chem. (side group react w/backbone?)
need 100% substitution?